

# Agricultural Nitrogen Control Practices and Options



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# My Assignment

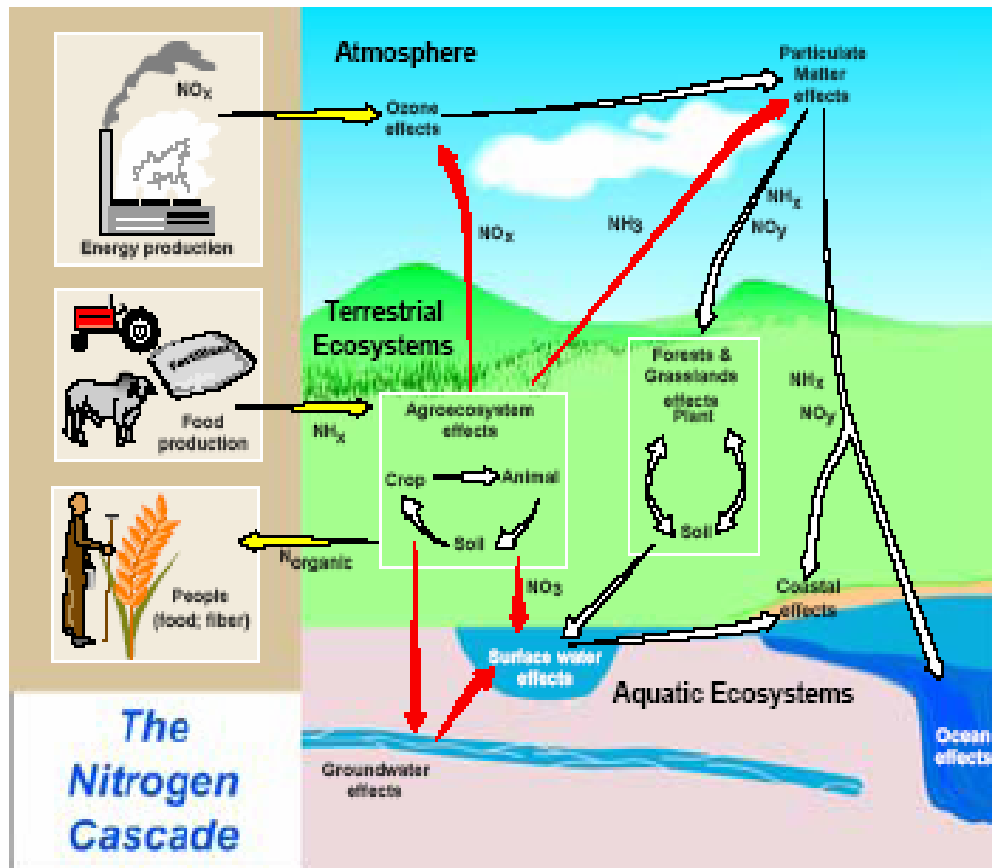
- State of the art for agricultural N emissions
- N emissions control practices
- Costs of control practices and technologies
- *All of that in 30 minutes*

## Summary

- ◆ Humans mobilize ~50% more Nr than natural terrestrial ecosystems.
  - *Food production accounts for 75%*
- ◆ Nr is widely dispersed
  - *Atmospheric Nr emissions have increased 3-fold since 1860; NH<sub>3</sub> twice as important as NO<sub>x</sub>*
  - *Nr is accumulating.*

## Next Questions

- ◆ What are the consequences of Nr emissions on the atmosphere and ecosystems?
- ◆ What should/can society do to slow or reverse Nr accumulation?



*Agroecosystems do not retain N very efficiently*

Galloway et al., 2003

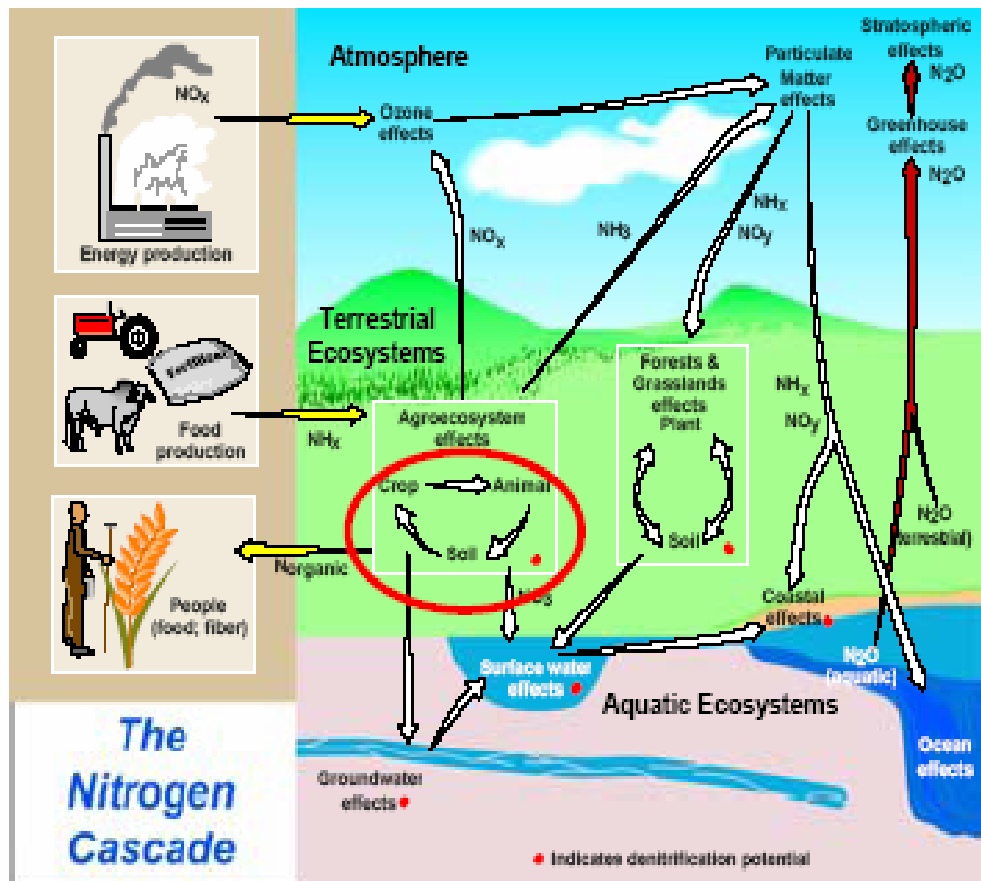
# Critical Control Points for Agricultural RNG Emissions

- **Crop Production**

- *Farm N balances*
- Inorganic N fertilizers
- Manure application
- Biomass decomposition

- **Livestock Production**

- *Farm N balances*
- Live-animal emissions
- Open-lot corral surfaces
- Ventilation exhausts
- Liquid manure and wastewater storages
- Composting facilities



**The Nitrogen Cascade**

*Let's talk about animal production*

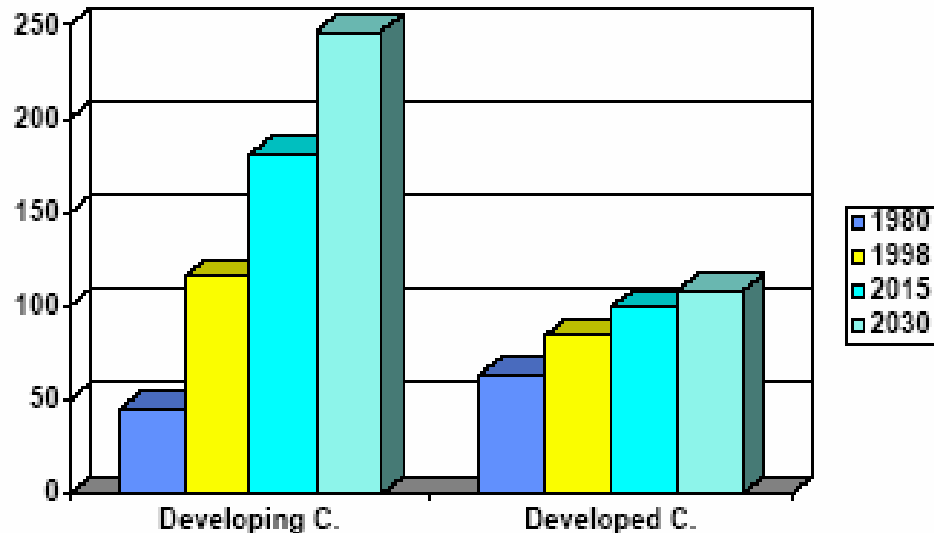
Galloway et al., 2003

James Galloway, "Human Alteration of the Nitrogen Cycle: Causes and Consequences," John Airy Symposium, Kansas City, MO, January 2006

# The changing livestock sector

## Increasing demand and production

Past and Projected Growth in Meat Production  
(million tons)



Henning Steinfeld, LEAD Project, FAO  
Livestock, Environment and Development

# Air Quality: Dust, Odor and Gases from Open-Lot Animal-Feeding Operations in the Southern Great Plains

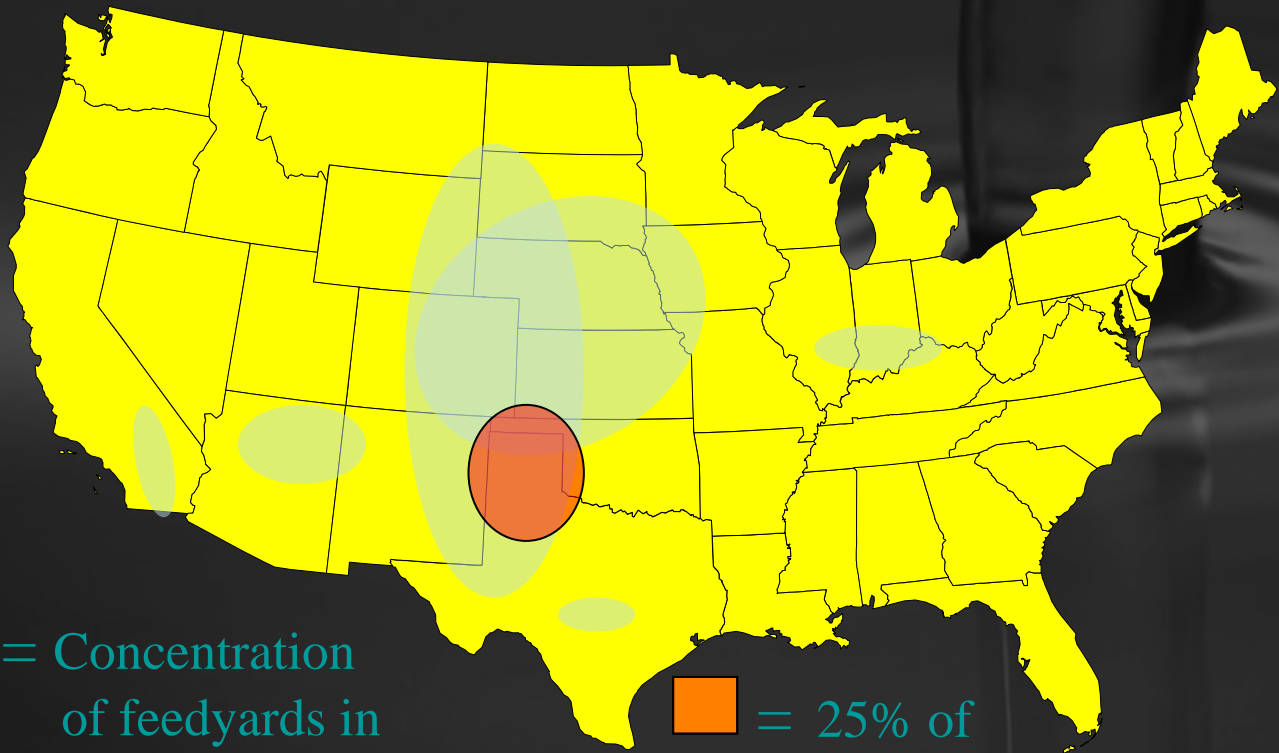
## Participants

Texas A&M Ag Program  
West Texas A&M  
University  
Kansas State University  
USDA Ag Research Service

## Major Objectives

1. Emissions Processes
2. Abatement Measures
3. Emission Factors
4. Health Effects
5. Technology Transfer





○ = Concentration  
of feedyards in  
U.S.

■ = 25% of  
nation's  
beef

# Fed-Cattle Industry in the US

# The Fed Cattle Industry in the United States

An aerial photograph of a large-scale cattle feedyard. The facility is divided into numerous rectangular pens, some of which are filled with dark-colored cattle. In the center of the complex, there are several large, light-colored buildings, likely for processing or storage, along with a large circular structure. The feedyard is situated in a rural area with green fields and a winding road or canal visible in the background.

- The trend to fewer, larger feedyards continues
- Nearly 60% of cattle are marketed from about 200 feedyards
- The number of cattle marketed from yards with fewer than 1,000 head has declined to under 3 million
- Average capacity in Texas High Plains: 40,000+

# Open-Lot Systems

- **Beef feedyards**
  - Animal spacing 75-250 ft<sup>2</sup>/hd
  - Excreted N 90% of N consumed in feed (Bierman et al., 1996)
- **Open-lot dairies**
  - Animal spacing 200-400+ ft<sup>2</sup>/hd
  - Excreted N 70% of N consumed in feed (Van Horn et al., 1996)



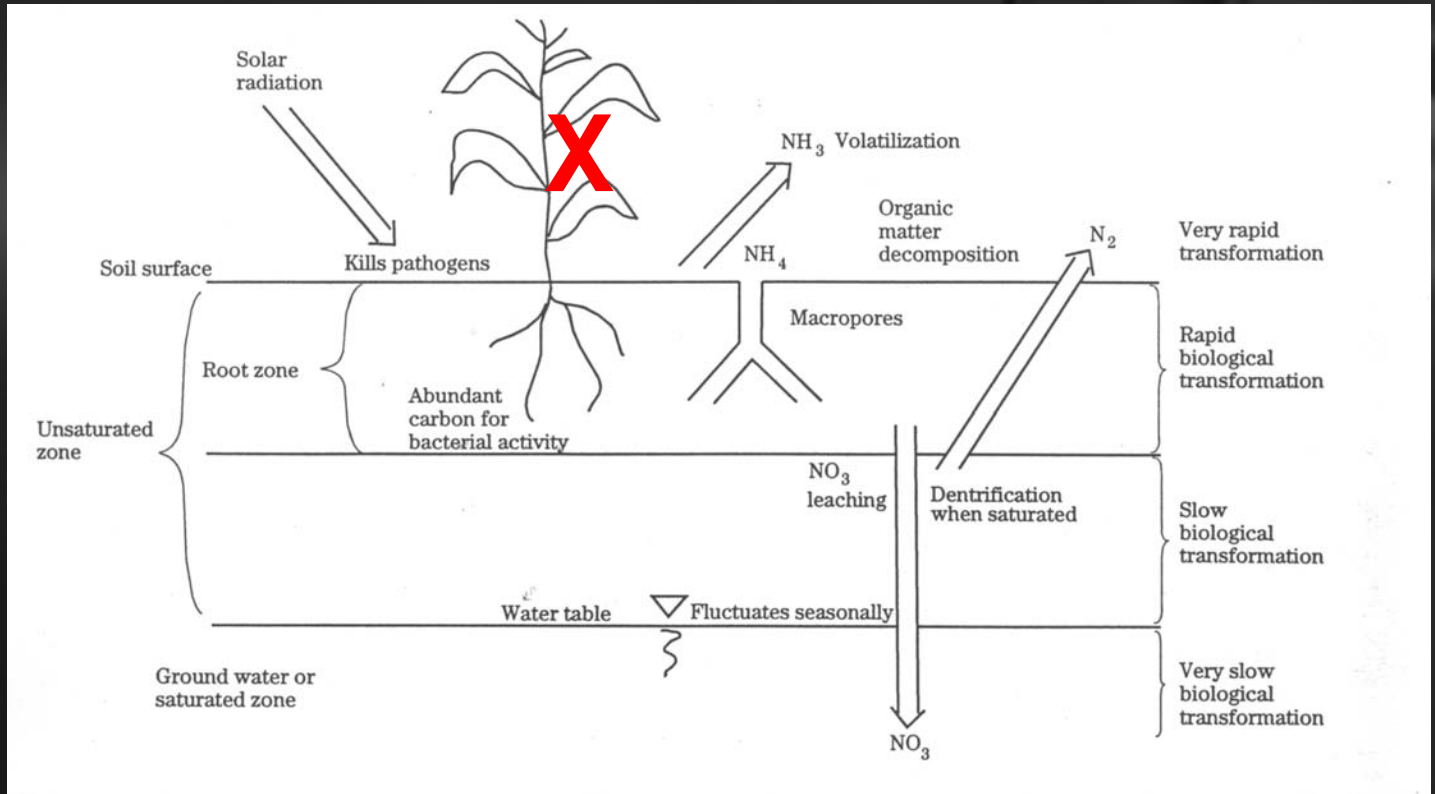
# Fate of Excreted N in Open-Lot Systems

- Collected in solid manure
  - Spread
  - Stored (stockpiles, mounds, other)
  - Composted and spread
- Remains on corral surface
  - Stable if it remains dry
  - Runs off into holding pond
- Volatilized as  $\text{NH}_3(\text{g})$  directly
  - Increases with wet/dry cycling

# NH<sub>3</sub> Loss: Open Lots vs. Ponds

- **Open lots**
  - Large area source, 2-9 acres per 1,000 head capacity
  - Variable emissions driven by wet/dry cycles, short-term temperature fluctuations
- **Lagoons and holding ponds**
  - Much smaller area source, 1-10 acres *total*
  - Seasonal temperature fluctuations
  - Continuous releases; f(temp, wind speed, RH)

# N Transformations on FY Surfaces



# NH<sub>3</sub> Concentrations Near Alberta Feedyards

- **Alberta Environment (2000)**
  - One-hour average concentrations
  - Up to ~800 μg m<sup>-3</sup> NH<sub>3</sub>-N
- **McGinn *et al.* (2003)**
  - Daily averages of 5-minute concentrations
  - Two highest values on days of lowest wind speeds
  - Up to ~1,500 μg m<sup>-3</sup> NH<sub>3</sub>-N

- **NH<sub>3</sub> presents steep challenges because of its:**
  - High reactivity with anions and surfaces
  - High aqueous solubility
    - Deposition
    - Condensation
  - Kinetically limited redox pathways w/NO<sub>x</sub> species
  - Numerous pools and pathways in real systems
  - Sensitivity to pH
- **Accounting for all of those factors in a single measurement scheme is complicated**
- **Uncertainty analysis assumes all sources of bias (systematic error) have been eliminated**



A dark, grayscale background featuring a glass of water with a ripple effect on the surface. The text is overlaid on this background.

# NH<sub>3</sub> Flux Estimates by 5 Methods

Courtesy N. A. Cole and R.  
Todd (2006)

# The Holy Grail

A range of emission factors that expresses the most probable, scientifically justifiable, annualized,  $\text{NH}_3$  emission flux from feedyards and dairies as a function of herd size, stocking density or other appropriate measure of capacity or throughput

# Where We Are Today

- There are dozens of different ways of estimating  $\text{NH}_3$  flux from an open-lot AFO
- Today, we consider results from several of them
- Getting at the *true flux* requires a convergence of results from independent methods, but even that's not enough

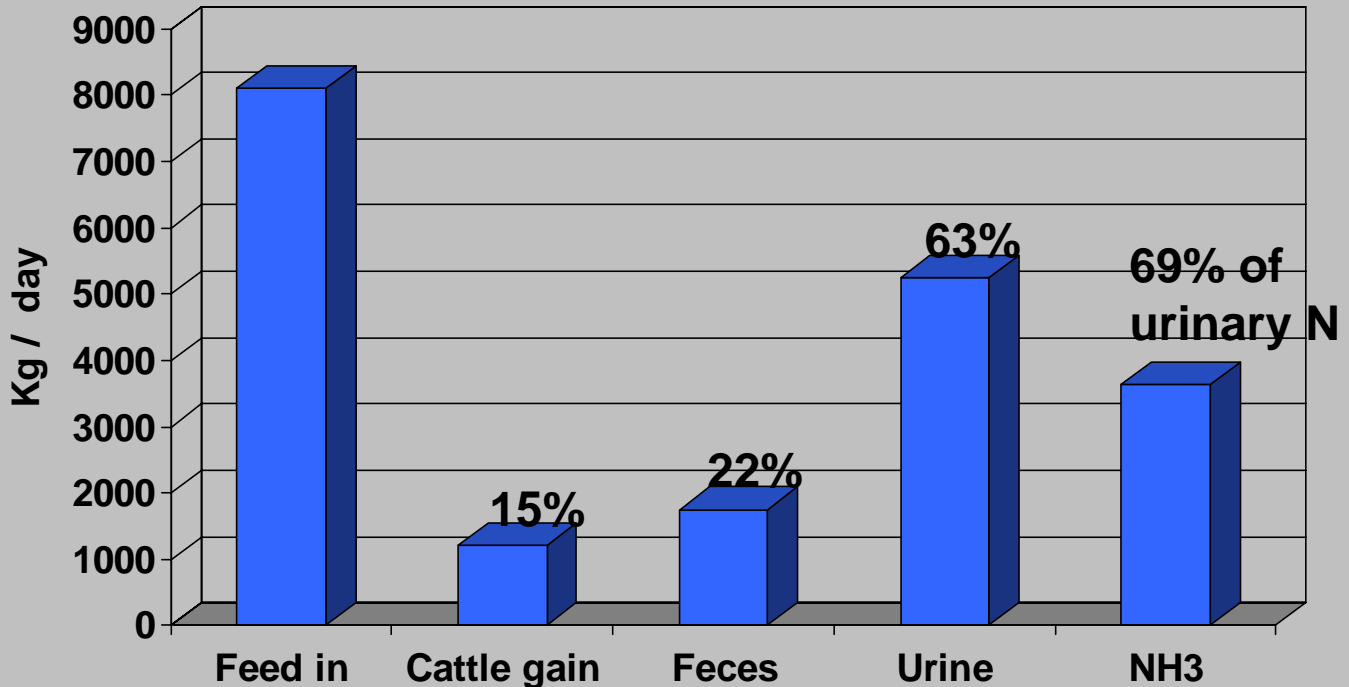
# Available Methods

- Envelope approaches
  - Mass balance
  - Nutrient ratio (N:P)
- Direct approaches
  - Surface isolation flux chambers
  - Wind tunnels
  - Eddy covariance
- Dispersion/box models
  - Gaussian (ISCST, AERMOD)
  - Lagrangian stochastic – backward, forward
  - Integrated horizontal flux (IHF)
  - Flux-gradient
  - Box

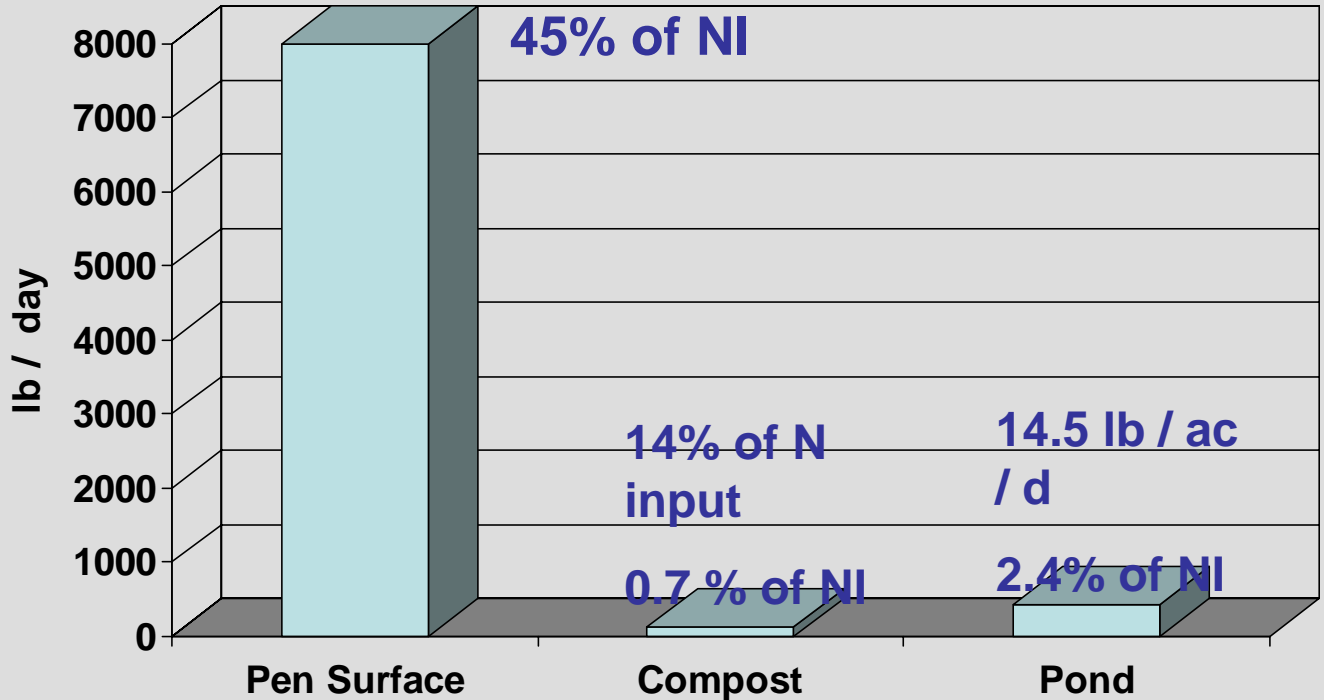
# Mass Balance Equation for Open-Lot AFOs

<b>Excr. Coef.</b>	<b>Total N Intake</b>		
	<b>Feed</b>	<b>Water</b>	
$M_{AT} = 1.21(1 - \gamma_{RN}) \left\{ \left[ \frac{C_{CP} \cdot M_{DMI}}{6.25} \right] + \left[ \frac{M_W \cdot C_{WN}}{10^6} \right] \right\} \{ \gamma_{EUN} \gamma_{UNV} + (1 - \gamma_{EUN}) \gamma_{FNV} \}$			
<b>Total N Excretion</b>			<b>Partitioning and Volatilization Coefficients</b>

# Feedyard N Balance



# Daily Volatile N Losses



Method	Beef	Dairy	Comments
	% of Fed N		
N Balance	44	<80	Uncertainty analysis nearly complete (beef)
N:P Ratio	48		Varies from 20-51% depending on source material (fresh manure, pen surface, compost)
Flux Chamber	18	3 (OL) – 5 (FS)	Herds are ~15% dry cows, ~85% lactating; excreted N is 79% of fed N
Flux-Gradient	43		Uncertainty analysis underway
bLS/OPL	41		Uses open-path lasers to measure N
Box Model	31-55		





CALVIN!

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# Urine-Spot $\text{NH}_3$ Emissions

- The vast majority of  $\text{NH}_3$  emissions comes from urine spots
- Surface chemistry changes rapidly
- Accurate measurements of  $\text{NH}_3$  (and  $\text{CH}_4$ ,  $\text{NO}_x$ ) flux are needed to develop appropriate models and make valid treatment comparisons

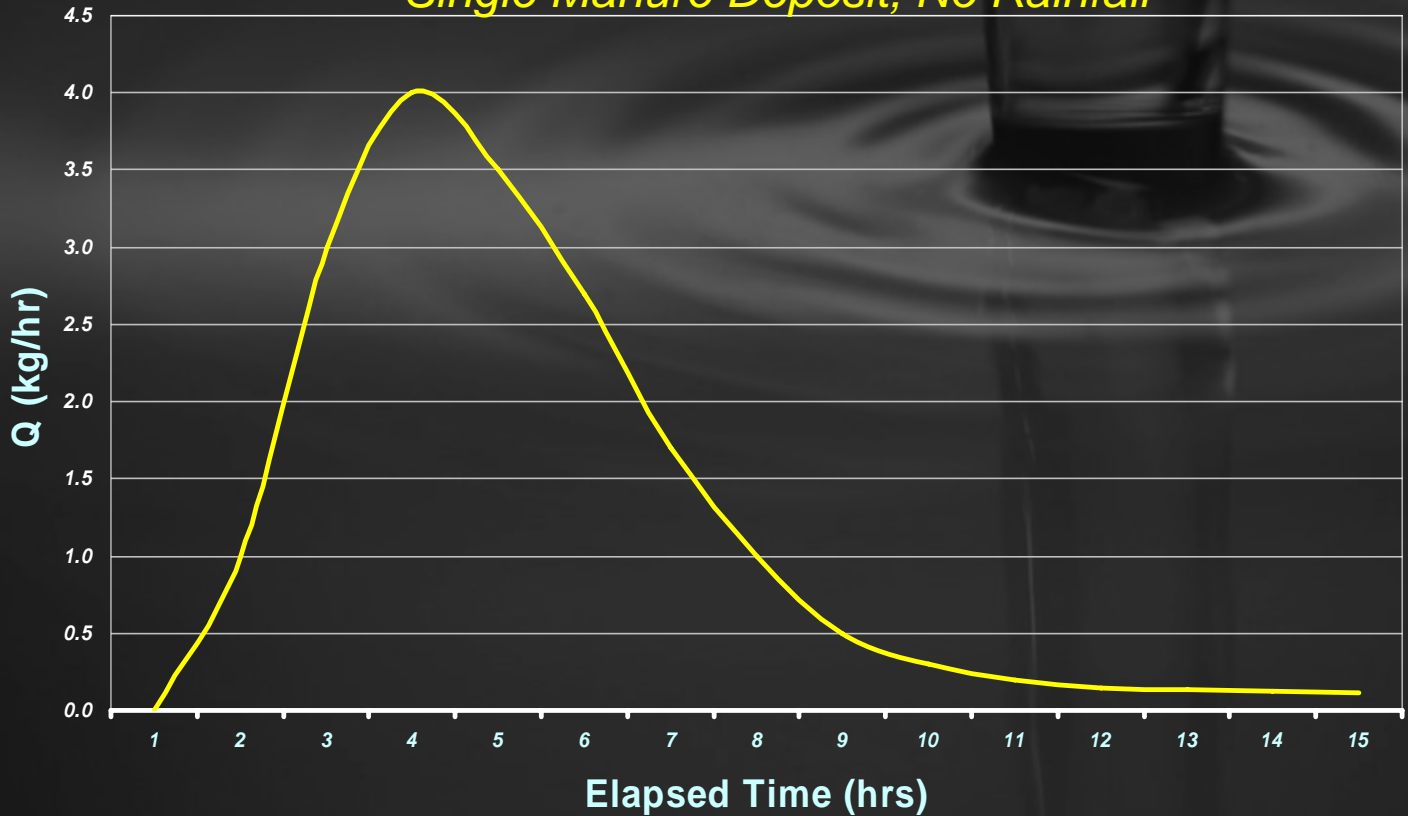
# Open-Lot $\text{NH}_3$ Flux: Drivers

- Wet/dry cycles
- Low C:N ratio of manure
- Favorable pH ( $>7.0$ )
- Enzyme-mediated hydrolysis of urea
- $\text{NH}_4^+$  highly soluble, mobile



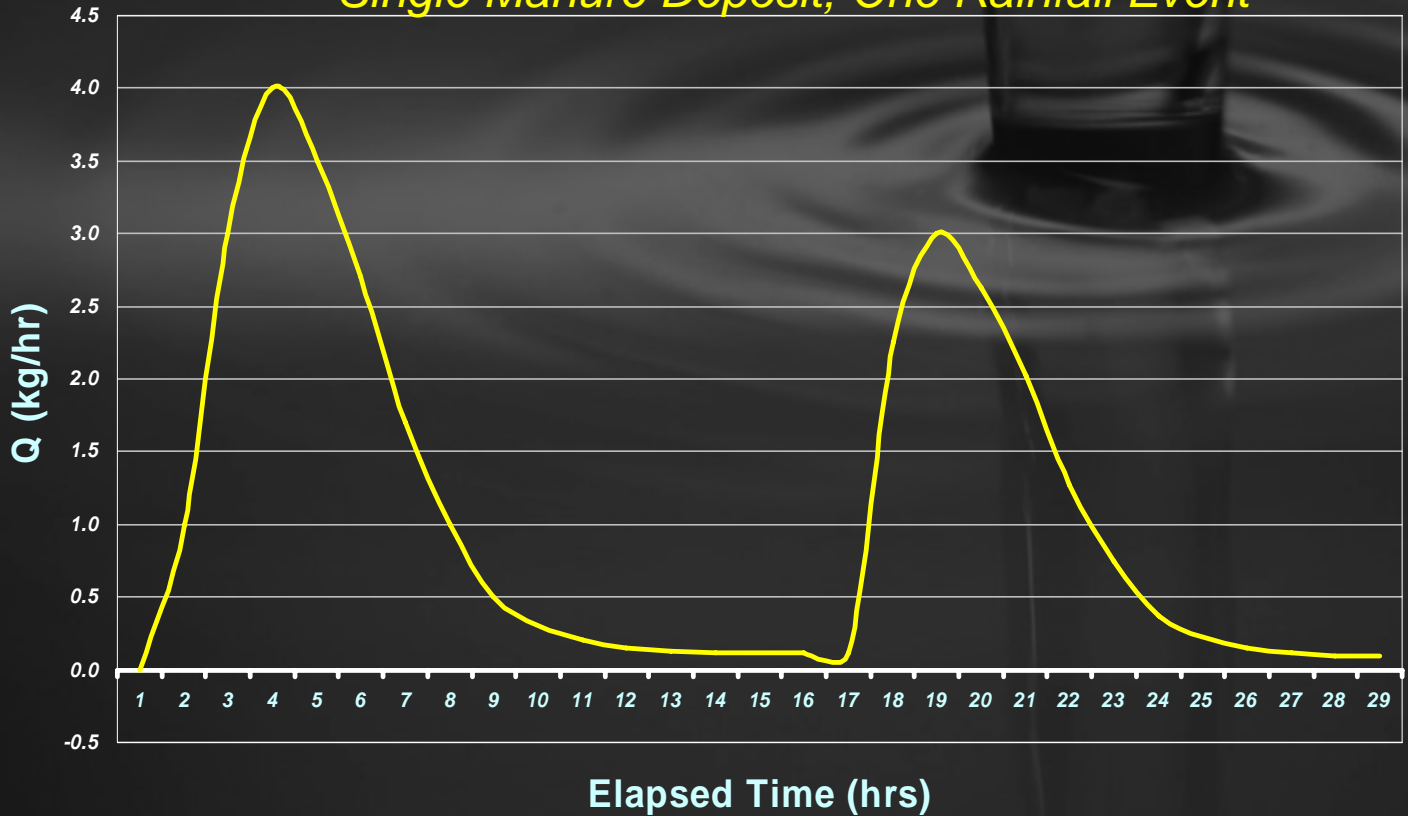
## Volatilization Rate vs. Time

*Single Manure Deposit; No Rainfall*



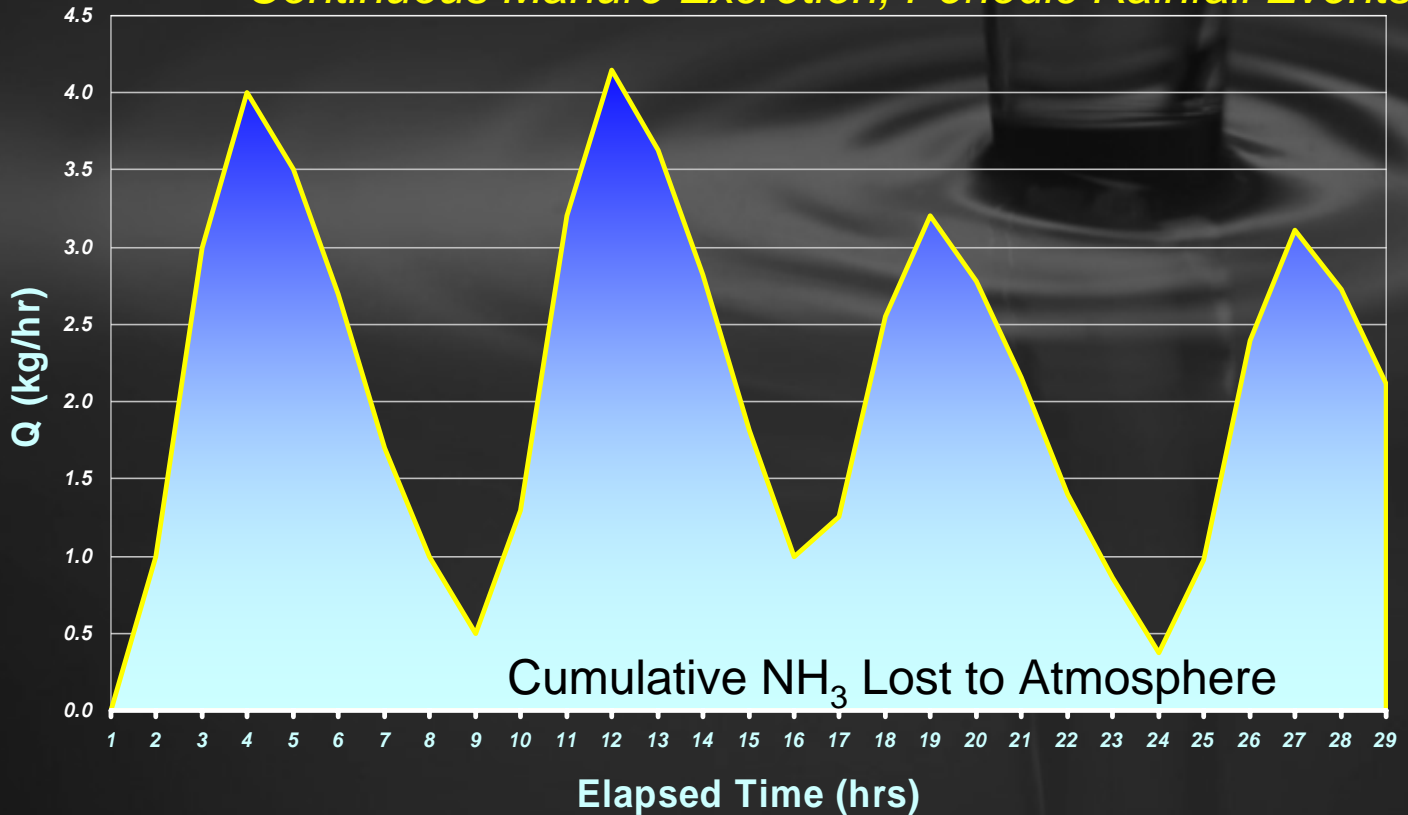
# Volatilization Rate vs. Time

## *Single Manure Deposit; One Rainfall Event*



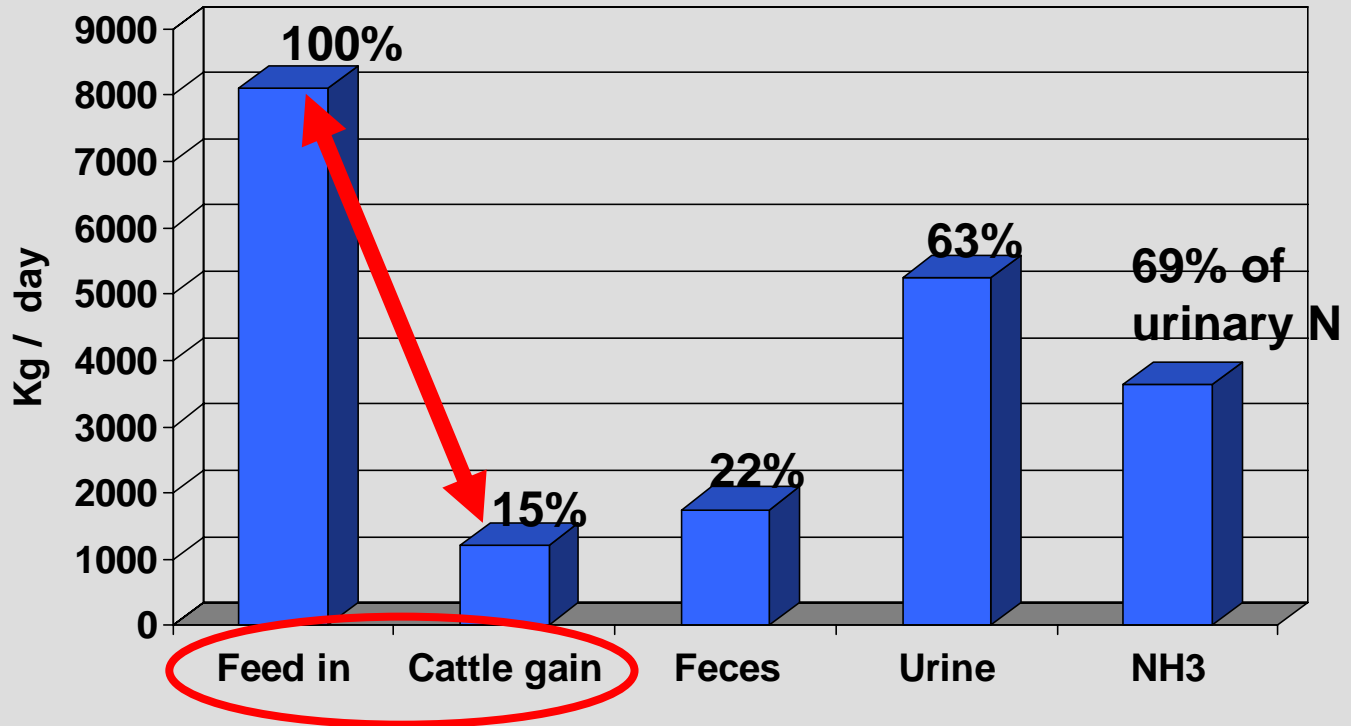
# Volatilization Rate vs. Time

*Continuous Manure Excretion; Periodic Rainfall Events*



Cumulative  $\text{NH}_3$  Lost to Atmosphere

# Feedyard N Balance

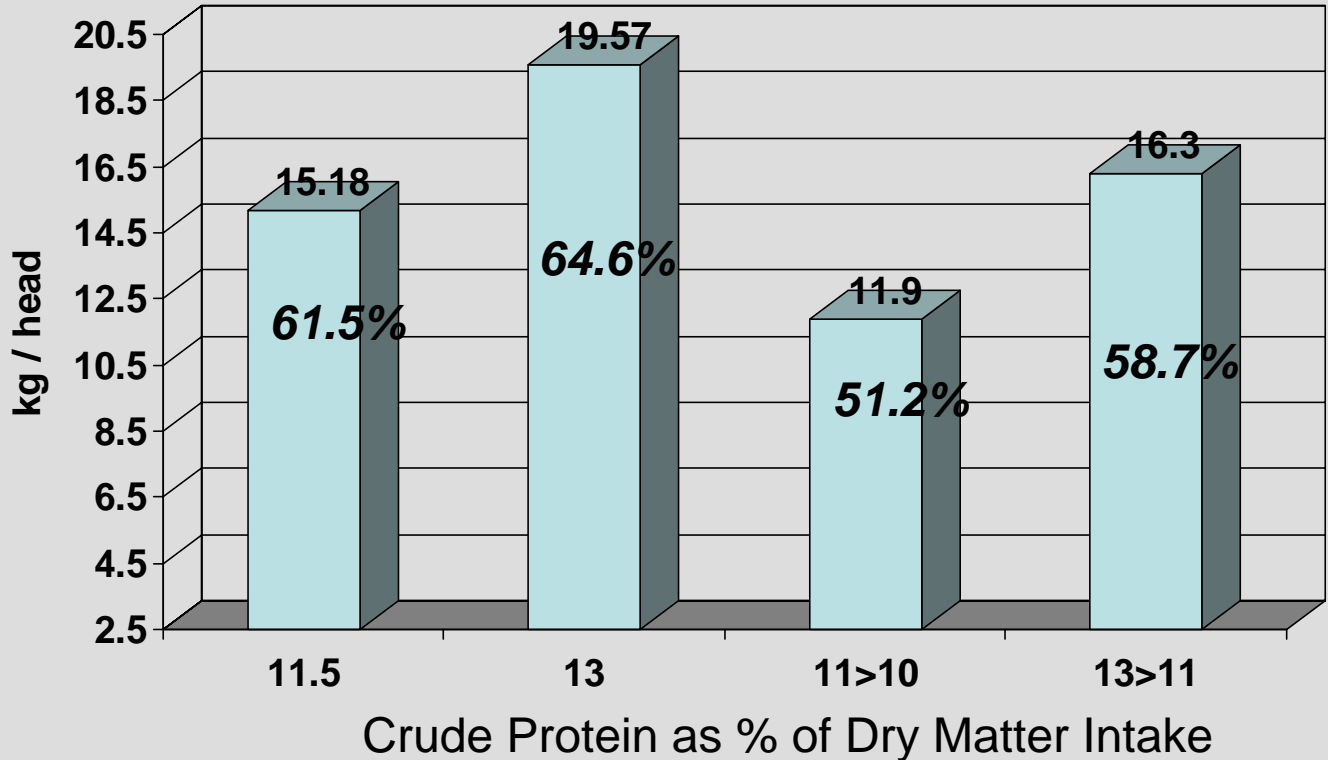




# Reducing Open-Lot $\text{NH}_3$ Flux

- Wet/dry cycles: *Stop  $\text{H}_2\text{O}$  applications, improve corral drainage*
- Low C:N ratio of manure: *Add carbonaceous bedding, mulch or liquid source of organic C (e. g., humates)*
- Favorable pH ( $>7.0$ ): *Alum or other*
- Enzyme-mediated hydrolysis of urea: *Urease inhibitors*
- $\text{NH}_4^+$  highly soluble, mobile: *Add strong adsorption sites (e. g., clinoptilite)*

# Effects of Phase Feeding on N Volatilization Losses



# Surface Amendments

- Shi et al. (2001) – *in vitro* evaluations of simulated feedyard surfaces
  - $\text{Al}_2(\text{SO}_4)_3$  – lowers manure pH
  - NBPT – suppression of urea hydrolysis to  $\text{NH}_4^+$
  - $\text{CaCl}_2$  – cation exchange
  - Humate (black and brown) – increase C:N ratio
- Measured cumulative loss over 21 days
- Incremental benefit computed as equivalent N fertilizer maintained in manure; rises and falls with NG/anhydrous prices
- Does not factor in the presence of carbonaceous bedding *as is common in Alberta feedyards*

# Results of Shi *et al.* (2001)

- Alum: 92% reduction at 4,500 kg/ha; B/C=0.17
- CaCl<sub>2</sub>: 71% reduction at 4,500 kg/ha; B/C=0.16
- ***NBPT: 65% reduction at 1 kg/ha; B/C=1.75***
- Humates: 65% reduction at 9,000 kg/ha;  
B/C=0.04

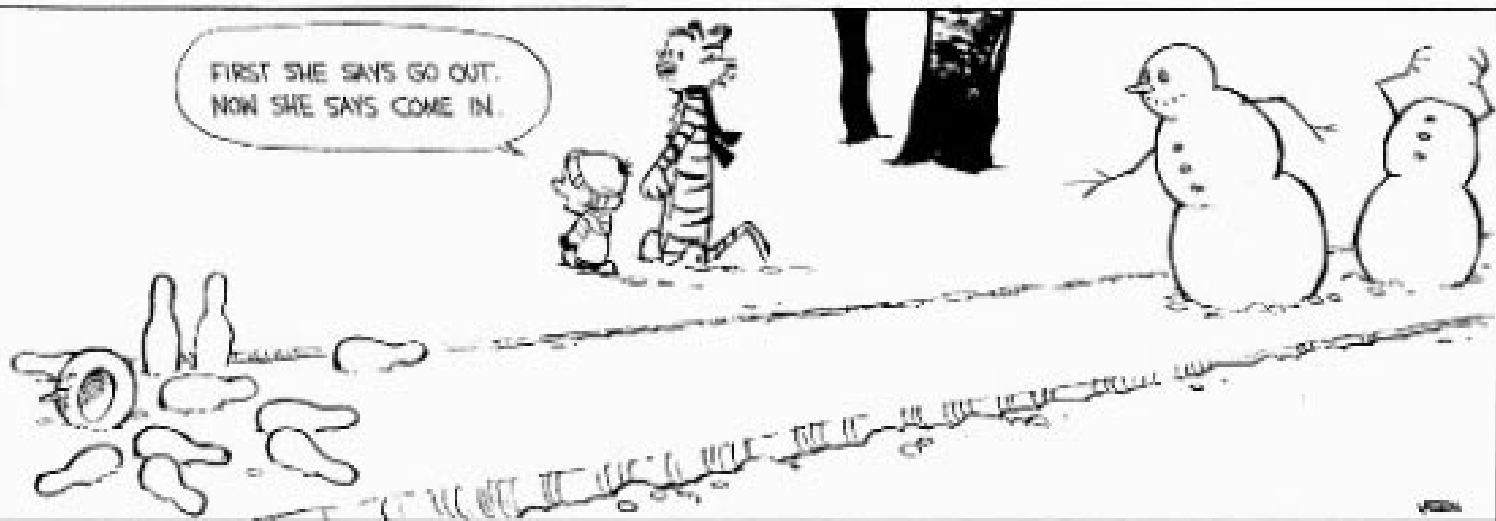
# Surface Amendments

- Replicating NBPT success outside the laboratory has been unsuccessful so far
- Keeping N as urea in manure surface would increase N pool and require increasing application rates over time
- Urea in solid manure can reduce  $\text{NO}_x$  emissions from coal-fired power plants during reburn

# Reducing Open-Lot $\text{NH}_3$ Losses

- Wet/dry cycles: *Stop  $\text{H}_2\text{O}$  applications, improve corral drainage*
- Low C:N ratio of manure: *Add carbonaceous bedding, liquid C source (humates; dilute beet extract?) or mulch*
- Favorable pH ( $>7.0$ ): *Alum or other*
- Enzyme-mediated hydrolysis of urea: *Urease inhibitors*
- $\text{NH}_4^+$  highly soluble, mobile: *Add strong adsorption sites (e. g., clinoptilite)*
- Extensive area source: *Manure harvesting*

FIRST SHE SAYS GO OUT.  
NOW SHE SAYS COME IN.









# A Reminder to Alberta's Policymakers

- We can design innovative stuff...
- ... but *can we afford it* at current levels of energy use?
- What about at *future* levels?

